

PROCESS FOR PRESERVATION OF CELLULOSIC MATERIALS

The present invention concerns a process for conservation of a cellulosic material.

The process is useful notably for conservation of valuable documents on cellulosic support such as books, manuscripts or works of art, clothes, flags and
5 the like.

During the past 150 years, archives and libraries have struggled to prevent the aging of paper, i.e. notably yellowing and embrittlement of paper in documents and books. Many treatments to avoid or stop this aging have been proposed. The primary goals of this treatment are either to transform the paper
10 into another, more stable medium or stabilise the paper against degradation processes.

Above documents are subject to deterioration notably because of the acid content in most of the cellulosic materials. Another serious problem encountered in the conservation of books and documents made of paper stems from the
15 readily oxidizable nature of the material. Oxidation not only results in the discolouration of paper but also a considerable mechanical weakening of the text's support. Such discolouration is favoured by light radiation, which constitutes
20 other significant source of degradation in paper. The auto-oxidizing mechanism of degradation which occurs in cellulose during the ageing process (in atmospheric oxygen) is also important in library conservation.

Canadian patent application CA-A-2142195 describes paper deacidification compositions containing methoxymagnesium methylcarbonate or ethoxymagnesium ethylcarbonate in a solvent diluted with a hydrochlorofluorocarbon or a hydrofluorocarbon.

25 This and other known conservation techniques are not entirely satisfactory as they may cause harm to inks and colours, thus giving rise to problems like ink or colour bleeding, formation of Newton rings and/or white deposits.

It was desirable to find a conservation process, notably a deacidification process which is effective and does not display the problems above, or which at
30 least allows to minimise said problems.

The invention concerns in consequence a process for conservation of a cellulosic material, comprising a treatment wherein said cellulosic material is

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contacted with a stabilising agent, and at least one of the cellulosic material or the stabilising agent is cooled before and/or during the treatment.

The cellulosic material is generally a printed and/or coloured cellulosic material.

5 The process according to the invention is compatible with the presence of other materials such as in particular leather used e.g. in book covers.

In the process according to the invention, at least one of the cellulosic material or the stabilising agent is generally cooled to a temperature which is less than 20°C. Often this temperature is equal to or less than 10°C. More often, this 10 temperature is equal to or less than 0°C. Preferably, this temperature is equal to or less than -10°C. A temperature of about -15°C is particularly preferred.

In the process according to the invention, at least one of the cellulosic material or the stabilising agent is generally cooled to a temperature which is equal to or higher than -50°C. Often, this temperature is equal to or higher than -30°C. A temperature equal to or higher than -25°C is particularly suitable. 15 Preferably, the temperature is equal to or higher than -20°C.

In a first preferred embodiment, at least the cellulosic material is cooled.

In a second embodiment, only the stabilising agent is cooled. If the 20 stabilising agent is a composition of several constituents, one or more of the constituents can eventually be cooled before forming the composition.

In a third, preferred, embodiment the treatment is carried out at the abovementioned temperature. That means that the system consisting of cellulosic material and stabilising agent is brought to these temperatures. Generally, in this case, the system is kept at that temperature for a time sufficient 25 to ensure the stabilising effect.

It is advantageous, in particular in the third embodiment, to ensure that the cellulosic material and the stabilising agent have substantially the same temperature as they are being contacted. Good results have been achieved with a 30 temperature difference between the cellulosic material and the stabilising agent of less than 20°C. Preferably, this temperature difference is less than 10°C and more preferably, it is less than 5°C. Most preferably, the temperature difference is about 1°C or less.

The process according to the invention is often carried out at a pressure at which the stabilising agent is in the liquid phase, being understood that 35 suspensions, dispersions and slurries are also considered as liquid phases. A solution is preferred when the stabilising agent is in the liquid phase.

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Alternatively, the process according to the invention can be carried out with a stabilising agent which is in the vapour phase under the treatment conditions.

Typical working pressures in the process according to the invention range from 1 to 10 bar.

5 In the process according to the invention, the duration of the treatment is generally at most 50 hours, preferably at most 10 hours. The duration of the treatment is more preferably less than or equal to about 3 hours. A duration of the treatment equal to or less than 1 hour, for example 30-45 min is more particularly preferred. However, treatment times of more than or equal to about 10 2 hours give also particularly good results.

It has been found, surprisingly, that the process according to the invention allows for short treatment times in spite of cooling stabilising agent and/or cellulosic material.

15 In a particular embodiment of the process according to the invention, the cellulosic material is not dried before the treatment.

In another particular embodiment of the process according to the invention, the cellulosic material is slightly dried before the treatment so that it loses about 1 – 2% of moisture content by weight relative to the total weight of the cellulosic material before treatment. Such slight drying can be brought about by gentle 20 heating of the cellulosic material to 30 to 50°C, preferably about 40°C.

A first particular way of carrying out the process according to the invention comprises :

- (a) cooling the stabilising agent;
- (b) contacting the cellulosic material and the stabilising agent cooled in step (a), 25 preferably in a treatment chamber which has optionally been cooled before introducing the cellulosic material;
- (c) optionally, separating excess quantities of stabilising agent or constituents of the stabilising agent from the cellulosic material;
- (d) optionally, recovering excess quantities or constituents separated in step (c).

30 A second particular way of carrying out the process according to the invention comprises :

- (a) providing a treatment chamber equipped with a cooling device, which treatment chamber is cooled down before the treatment;
- (b) introducing cooled cellulosic material into the treatment chamber;
- 35 (c) supplying the optionally cooled stabilising agent to said treatment chamber so as to contact the stabilising agent with the cellulosic material;

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(d) optionally, separating excess quantities of stabilising agent or constituents of the stabilising agent from the cellulosic material;

(e) optionally, recovering excess quantities or constituents separated in step (d).

A third particular way of carrying out the process according to the invention comprises :

(a) cooling the cellulosic material and optionally cooling the stabilising agent;

(b) contacting the cellulosic material cooled in step (a) and the stabilising agent optionally cooled in step (a) in a treatment chamber;

(c) optionally, separating excess quantities of stabilising agent or constituents of the stabilising agent from the cellulosic material;

(d) optionally, recovering excess quantities or constituents separated in step (c).

In the third particular way of carrying out the process according to the invention, at least the cellulosic material, but optionally also the stabilizing agent are cooled before bringing them into contact. It is preferred to cool both cellulosic material and stabilizing agent before contacting them. The treatment chamber may be of a simple type which comprises no cooling equipment or in which cooling equipment is not actuated.

In the process according to the invention, the stabilising agent is often selected from fibre strengtheners, sizing agents, antioxidants, biocides and/or deacidification reagents.

Among fibre strengtheners and sizing agents, cellulose derivatives and/or natural or non-natural polymers can suitably used. Specific examples of cellulose derivatives are selected from carboxymethylcellulose, methylcellulose, ethylcellulose and cellulose ethers. Specific examples of non-natural polymers are selected from styrene/acrylate copolymers, polyurethanes and polyamides. A specific natural polymer is starch.

Among antioxidants, derivatives of oxidizable arylcarboxylic acids, in particular hydroxyarylcarboxylic acids can suitably be used. Esters of arylcarboxylic acids are preferred. In this case the ester is generally an alkyl ester, preferably an alkyl ester derived from a linear, branched or cyclic alkyl radical having generally from 1 to 10, preferably from 1 to 4 carbon atoms, such as a methyl, ethyl, propyl, isopropyl or butyl radicals, or their mixtures.

In a particularly preferred embodiment a derivative, in particular an alkyl ester of the p-hydroxibenzoic acid is used which is preferably selected among methyl p-hydroxibenzoate, ethyl p-hydroxibenzoate, propyl p-hydroxibenzoate, butyl p-hydroxibenzoate and their blends.

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Preferred biocides correspond to the antioxidants described herebefore.

In the process according to the invention, the stabilising agent is preferably a deacidification agent.

5 In this case, the deacidification agent generally comprises a base which can be selected for example from basic metal derivatives, from organometallic derivatives or from alkaline monomers such as acrylates.

10 Examples of particular basic metal derivatives suitable for use in the present invention include the oxides, hydroxides, carbonates and bicarbonates of zinc and metals in Group I, II and IV of the Periodic Table. According to certain embodiments, the bases are preferably oxides, hydroxides, carbonates and bicarbonates of magnesium, titanium, calcium, sodium, potassium, zinc, or combinations of two or more thereof. Examples of agents include magnesium carbonate, magnesium bicarbonate, magnesium oxide, magnesium methyl carbonate, titanium alcoholates, calcium oxide, sodium hydroxide, potassium 15 hydroxide, calcium hydroxide, zinc carbonate, zinc bicarbonate, zinc oxide, and combinations of two or more thereof.

20 Basic alkaline earth metal derivatives in particular magnesium compounds or salts are preferred. Most preferred bases are selected from magnesium alkoxycarbonates such as carbonates derived from magnesiumdimethanolate, magnesiumdiethanolate and in particular magnesiumdipropanolate.

25 The deacidification agent may optionally comprise a solvent selected for example from alcohols, in particular having 1 to 4 carbon atoms, and non-halogenated or halogenated hydrocarbon solvents or ethers. Such solvents include for example methanol, ethanol, isopropanol, n-propanol, isobutanol, propane, butanes, pentanes, iso hexanes, heptanes, alkylsiloxanes, HFE-7100 and HFE-7200. n-propanol is preferred.

In another embodiment, the solvent comprises a fluorinated alcohol such as 2,2,3,3,3-pentafluoro-1-propanol.

30 In a preferred embodiment of the invention, the stabilising agent, in particular the deacidification agent comprises a hydrofluoroalkane. Suitable hydrofluoroalkanes generally comprise from 1 to 10, preferably from 2 to 6 carbon atoms.

35 Specific hydrofluoroalkanes are selected from difluoromethane (HFC-32), 1,1-difluoroethane (HFC-152a), 1,1,1-trifluoroethane (HFC-143a), 1,1,1,2-tetrafluoroethane (HFC-134a), 1,1,1,3,3-pentafluoropropane (HFC-245fa), 1,1,1,3,3,3-hexafluoropropane (HFC-236fa),

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1,1,1,2,3,3,3-heptafluoropropane (HFC-227ea), 1,1,1,3,3-pentafluorobutane (HFC-365mfc) and 1,1,1,2,3,4,4,5,5,5-decafluoropentane (HFC-43-10mee). HFC-227ea, HFC-134a and their mixtures are preferred.

A surfactant may optionally be present, in particular if no solvent is used.

5 The moisture content of optional solvent and hydrofluoroalkane is generally less than 1000 mg/kg. This content can also be below 200 or 100 mg/kg.

In a most preferred embodiment, the deacidification agent is a composition of magnesium propylcarbonate, propanol, in particular n-propanol, and
10 HFC-227ea and/or HFC-134a. This composition and its manufacture is described in EP-11111128 whose content is incorporated by reference in the present application. This particular deacidification agent is generally in the form of a solution and is comprised of carbonated magnesium di-n-propylate preferably in n-propanol, and an HFC diluent selected from 1,1,1,2-tetrafluoroethane (HFC-134a) and 1,1,1,2,3,3,3-heptafluoropropane (HFC-227).
15 Carbonated magnesium di-n-propylate is a solid with the formula $(CH_3CH_2CH_2O)_2MgOCO$. The concentration of carbonated magnesium di-n-propylate in n-propanol is preferably between 30 and 70% (weight by weight relative to the total weight of carbonated magnesium di-n-propylate and n-propanol). A composition comprising 1.5 – 2.0% (weight by weight relative to the total weight of the composition) of carbonated magnesium di-n-propylate, 3.0 - 4.0% of n-propanol and the remainder consisting essentially of HFC is more particularly preferred.
20

The invention is illustrated in a non-limitative manner by the examples
25 hereafter :

General remarks :

A set of books printed on acidic paper in the first half of the twentieth century and having handwritten ink notes written on several pages is selected due to the sensitivity of the mentioned inks to deacidification treatments.

30 Examples 1 to 3 were performed using a chamber with cooling equipment which was cooled during the contact of books and stabilizing agent.

Examples 4 to 6 were performed in a simple treatment chamber without cooling equipment.

35 Example 7 (comparison example) was performed without cooling book and stabilizing agent before and during contact.

Examples :

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Example 1 : Deacidification process performed in a cooled treatment chamber

In a treatment chamber, a book from the set of books mentioned under "General remarks" above was cooled to a temperature of -15°C. A deacidification reagent corresponding to the preferred embodiment described on p. 7, lines 12 to 20 was diluted with HFC-227ea and the diluted reagent was cooled to -15°C. It was then introduced into the treatment chamber while controlling the temperature so as to make sure that the difference between the deacidification reagent and the book did not exceed about 3°C. The system reagent/book was kept together for 10 minutes at -15°C for impregnation, then the chamber was drained for 30 minutes and then the remaining volatiles were removed by evaporation for 1.5 hours.

The treated book was effectively deacidified and showed no ink bleeding at the handwritten ink notes.

Example 2

Example 1 was repeated with the same type of book and the same deacidification reagent but the temperatures were -20°C. The treated book was effectively deacidified and showed no ink bleeding at the handwritten ink notes.

Example 3

Example 1 was repeated with the same type of book and the same deacidification reagent but the temperatures were -25°C. The treated book was effectively deacidified and showed no ink bleeding at the handwritten ink notes.

Example 4 : Deacidification process performed in a treatment chamber without cooling equipment

A book of the set of books as mentioned under "General remarks" above was cooled to a temperature of -15 °C. A deacidification reagent corresponding to the preferred embodiment described on p. 7, lines 12 to 20 was diluted with HFC-227ea and the diluted reagent was cooled to -15°C. Both book and deacidification reagent were then introduced into a simple treatment chamber. The system reagent/book was kept together for 10 minutes for impregnation, then the chamber was drained for 30 minutes and then the remaining volatiles were removed by evaporation for 1.5 hours.

The treated book was effectively deacidified and showed no ink bleeding at the handwritten ink notes.

Example 5

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Example 4 was repeated with the same type of book and the same deacidification reagent but the temperatures were -20°C. The treated book was effectively deacidified and showed no ink bleeding at the handwritten ink notes.

Example 6

5 Example 4 was repeated with the same type of book and same deacidification reagent but the temperatures were -25°C. The treated book was effectively deacidified and showed no ink bleeding at the handwritten ink notes.

Of course, examples 4 to 6 as well could have been performed in a treatment chamber with cooling equipment kept switched off.

10 **Example 7 (comparison)**

Example 1 was repeated with the same type of book and the same deacidification reagent but without temperature control, at room temperature. The treated book was deacidified but showed considerable ink bleeding at the handwritten ink notes.